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## Key indicators

Single-crystal X-ray study  
 $T = 153\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.008\text{ \AA}$   
H-atom completeness 97%  
 $R$  factor = 0.050  
 $wR$  factor = 0.085  
Data-to-parameter ratio = 11.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

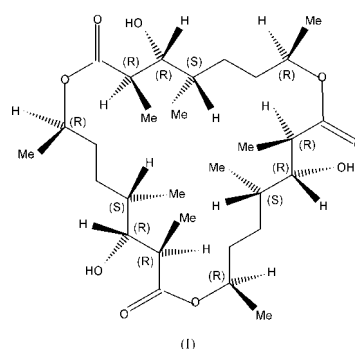
## Dasypogalactone–methanol–water (1/2/1)

In the crystal structure of the title compound, 4,12,20-trihydroxy-3,5,8,11,13,16,19,21,24-nonamethyl-1,9,17-trioxacyclotetracosane-2,10,18-trione–methanol–water (1/2/1),  $\text{C}_{30}\text{H}_{54}\text{O}_9 \cdot 2\text{CH}_4\text{O} \cdot \text{H}_2\text{O}$ , the molecules are packed parallel to [100], with various hydrogen bonds to trapped  $\text{CH}_3\text{OH}$  and  $\text{H}_2\text{O}$  solvent molecules.

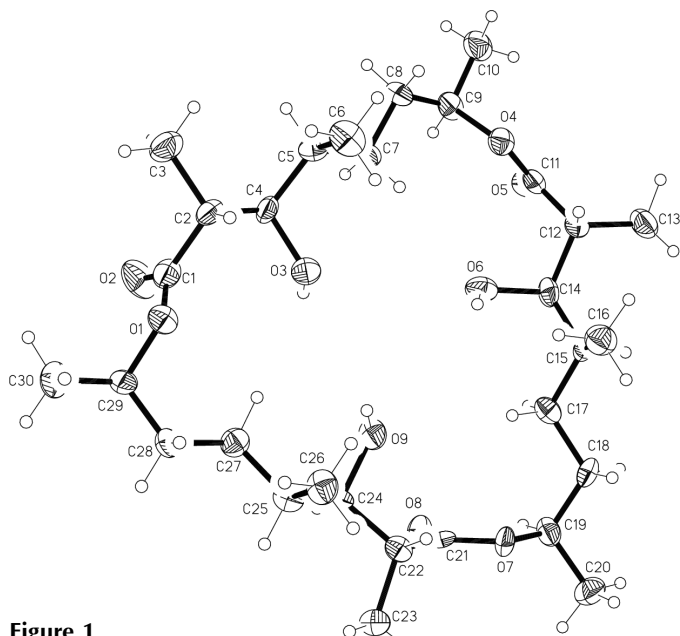
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## Comment

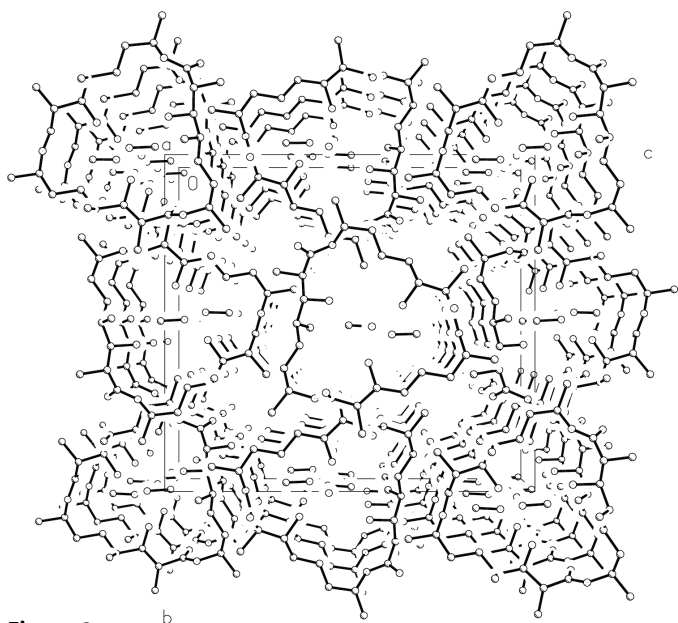
In an earlier study, we reported the isolation of dasypogalactone from the Indonesian lichen *Usnea Dasypoga* Rohl. and proposed its molecular structure from mass spectra and NMR data (Suwarso *et al.*, 1999). However, the configuration of dasypogalactone was not fully established. Now, we have succeeded in growing crystals, still of poor quality and scattering power [only 1961 intensities with  $I > 2\sigma(I)$ ], but at least suitable for a single-crystal X-ray analysis. It was not possible to determine the absolute configuration, but the former structure was confirmed with the correct relative configuration assigned as (2,3,7*R*\*)-(4*S*\*)-3,7-dihydroxy-2,4-dimethyloctanoic acid for the monomeric acid unit (see *Scheme*).



The molecule of the title compound, (I), exhibits non-crystallographic  $C_3$  symmetry, with the OH groups (O3, O6, O9; for numbering see Fig. 1) lying in the ring plane and directed towards the ring centre. Their intramolecular non-bonding distances are  $\text{O3} \cdots \text{O6} = 3.966(5)$ ,  $\text{O3} \cdots \text{O9} = 3.544(5)$  and  $\text{O6} \cdots \text{O9} = 3.978(5)$  Å. Carboxylic acid O atoms O2, O5 and O8 point to one side of the ring plane and the three methyl groups C6, C16 and C26 to the other. The three parts of the ring show almost equal bond geometries at the  $3\sigma$  level. In the crystal structure, the molecules are packed head-to-tail along [100] (Fig. 2), with a plane-to-plane distance of 8.898(1) Å. Enclosed solvent molecules, one water and two methanol per asymmetric unit, are linked to this host lattice by various hydrogen bonds. The MeOH solvent molecule 2 (C200/O200) is situated, sandwich-like, midway between two dasypogalactone rings (Fig. 3), with C200 and O200 4.390(7)

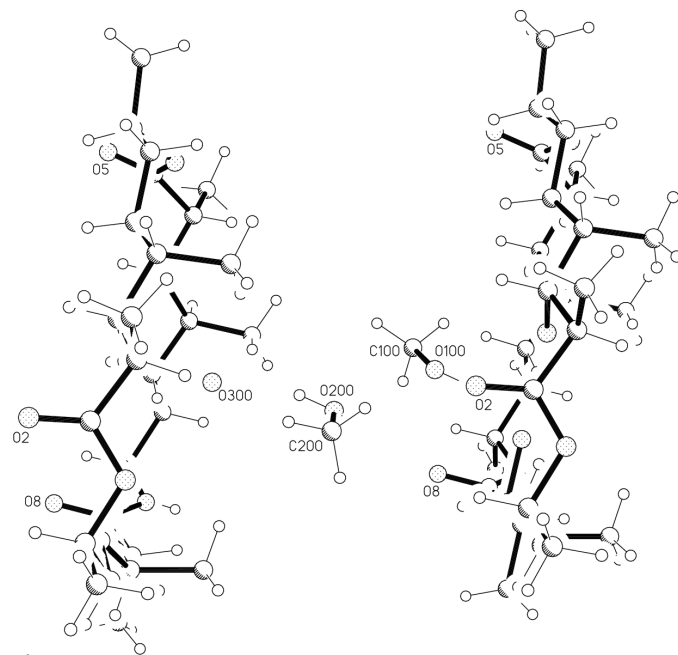


**Figure 1**  
The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level, and solvent molecules are omitted.



**Figure 2**  
Packing diagram, viewed along [100], with H atoms omitted.

and 4.327 (5) Å, respectively, above the plane defined by O3, O6 and O9. This solvent molecule shows an O200—H200···O300 hydrogen bond to the water atom O300, with H200···O300 = 1.84 Å and an angle of 160°, and a weak C200—H200E···O2 interaction (2.43 Å; 158°). Atoms C100 and O100 of methanol solvent molecule 1 lie 2.601 (7) and 2.181 (5) Å above the O3/O6/O9 plane. The corresponding hydrogen bonds are O100—H100···O9 (1.95 Å, 149°) and C100—H10D···O6 (2.37 Å, 167°). The water atom O300 is located 6.967 (4) Å above the O3/O6/O9 plane; additional hydrogen bonds are O6—H6···O300( $x + 1, y, z$ ) (2.08 Å,



**Figure 3**  
The position of the solvent molecules between two dasypogalactone rings.

154°) and O9—H9···O300( $x + 1, y, z$ ) (2.12 Å, 126°). It was not possible to locate the water H atoms. All the above values are normalized for C—H = 1.08 Å and O—H = 0.938 Å. A somewhat similar packing of C<sub>3</sub>-symmetric macrolides, with trapped water and methanol solvent molecules, was described recently (Burke & Zhao, 2000).

## Experimental

Isolation of the compound has already been described by Suwarso *et al.* (1999). The product was crystallized from a CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH solution.

### Crystal data

C<sub>30</sub>H<sub>54</sub>O<sub>9</sub>·2CH<sub>4</sub>O·H<sub>2</sub>O  
*M<sub>r</sub>* = 640.83  
 Orthorhombic, *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>  
*a* = 8.8977 (14) Å  
*b* = 19.524 (3) Å  
*c* = 21.445 (3) Å  
*V* = 3725.5 (10) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.143 Mg m<sup>-3</sup>

Mo *K*α radiation  
 Cell parameters from 713 reflections  
 $\theta$  = 2.5–12.7°  
 $\mu$  = 0.09 mm<sup>-1</sup>  
*T* = 153 (2) K  
 Block, colourless  
 0.32 × 0.25 × 0.15 mm

### Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2002)  
 $T_{\min}$  = 0.906,  $T_{\max}$  = 0.953  
 22975 measured reflections

4594 independent reflections  
 1359 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}}$  = 0.108  
 $\theta_{\text{max}}$  = 27.1°  
 $h$  = -11 → 8  
 $k$  = -24 → 24  
 $l$  = -27 → 26

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)]$  = 0.050  
 $wR(F^2)$  = 0.085  
 $S$  = 0.86  
 4594 reflections  
 413 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0003P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

**Table 1**  
Selected geometric parameters (Å, °).

|           |           |           |           |
|-----------|-----------|-----------|-----------|
| O1—C1     | 1.330 (7) | C4—C5     | 1.526 (7) |
| O1—C29    | 1.455 (6) | C5—C6     | 1.508 (7) |
| O2—C1     | 1.203 (7) | C5—C7     | 1.516 (7) |
| O3—C4     | 1.464 (5) | C7—C8     | 1.503 (6) |
| C1—C2     | 1.502 (8) | C8—C9     | 1.504 (7) |
| C2—C4     | 1.527 (6) | C9—C10    | 1.523 (6) |
| C2—C3     | 1.528 (6) |           |           |
| C1—O1—C29 | 117.2 (5) | C2—C4—C5  | 115.4 (5) |
| O2—C1—O1  | 124.4 (7) | C6—C5—C7  | 109.3 (6) |
| O2—C1—C2  | 122.7 (7) | C6—C5—C4  | 113.4 (5) |
| O1—C1—C2  | 112.8 (6) | C7—C5—C4  | 110.1 (5) |
| C1—C2—C4  | 108.8 (5) | C8—C7—C5  | 116.6 (5) |
| C1—C2—C3  | 111.0 (5) | C7—C8—C9  | 113.3 (5) |
| C4—C2—C3  | 111.7 (5) | O4—C9—C8  | 105.5 (5) |
| O3—C4—C2  | 106.1 (5) | O4—C9—C10 | 109.4 (5) |
| O3—C4—C5  | 109.8 (5) | C8—C9—C10 | 113.5 (5) |

H atoms were placed at calculated positions riding on the C or O atoms, with isotropic displacement parameters  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}(\text{CH}_3/\text{OH})$ . All  $\text{CH}_3$  and OH groups were allowed to rotate but

not to tip. The H atoms of the water molecule could not be located and were, therefore, not included in the refinement. The title compound crystallizes in the non-centrosymmetric space group  $P2_12_12_1$ ; however, in the absence of significant anomalous scattering effects, the Flack (1983) parameter is essentially meaningless. Accordingly, Friedel pairs were merged.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2002); program(s) used to solve structure: *SHELXTL* (Bruker, 2002); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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